## Characterizing Small Molecules in Biological Extracts using IntelliXtract Data Processing on High Resolution Accurate Mass Time-of-Flight Data

### INTRODUCTION

Data Acquisition The three phenotypes of Zucker rat were investigated by To detect small molecules in biological extracts, LC/MS LC-HRT analysis of plasma samples, import of the performed using high resolution time of flight with accurate Plasma samples from lean, fatty, and obese Zucker rats data to ACD/MS Manager software, and then mass analysis is a powerful technology. Together with were prepared and analyzed using LC-high resolution mass ACD/IntelliXtract COMPARE function processing. advanced data processing algorithms, accurate mass, and spectrometry (HRT). Samples were filtered (5000 MWCO, Common (similar/different) and unique components relative isotope abundance measurements, molecular Microcon), then diluted 5x and analyzed. Siderophores were were detected with the classification being examined as formulae can be determined and compounds identified. extracted from cell cultures (yersinia) and the isolates analyzed Compounds excreted or secreted by organisms can reveal an approach to distinction. Extracted [M+H]<sup>+</sup> ions could using an LC-HRT. Data were imported to ACD/MS Manager aspects of their normal metabolism or disease states. be further analyzed to determine molecular formula and v12.01 software using a new custom developed import filter Illustrative examples include profiling of plasma to investigate compounds consistent with some expected phospholipid for HRT data. Molecular formulae could be proposed and the the Zucker rat animal model of obesity/diabetes, and probing and carnitine classes, including carnitine and acyl ACD/IntelliXtract COMPARE function was used to process and carnitines C2:0, C3:0, and C4:0. of bacterial cultures for siderophores—compounds used to compare relevant regions of the rat sample chromatograms. scavenge trace metals. Fragmentation algorithms were used to look at in-source Formula generation was feasible since measured mass dissociation spectra acquired from yersinia samples. uncertainties of 1 ppm were offered by the LC-HRT

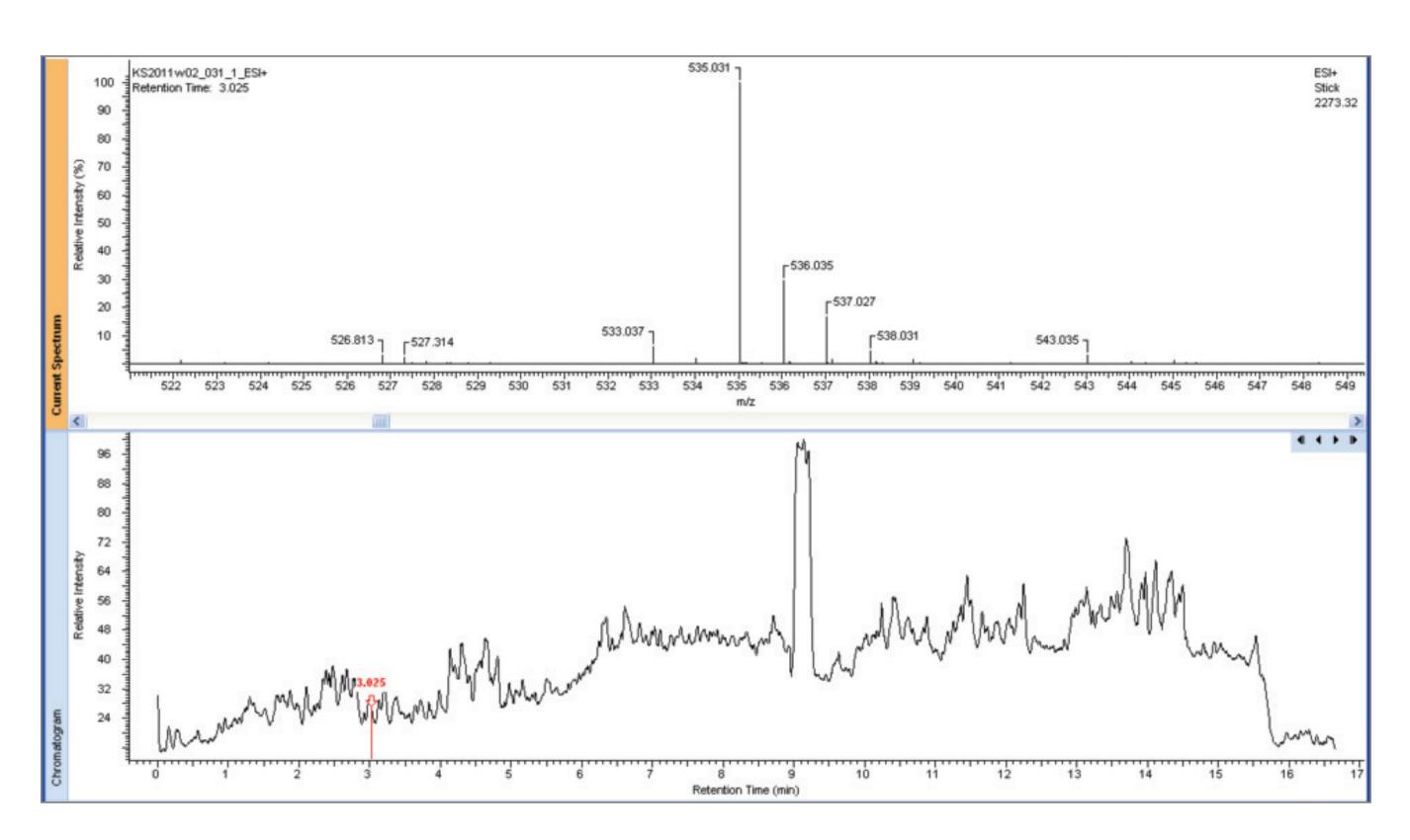


Figure 1. Spectrum of iron siderophore.



# METHODS

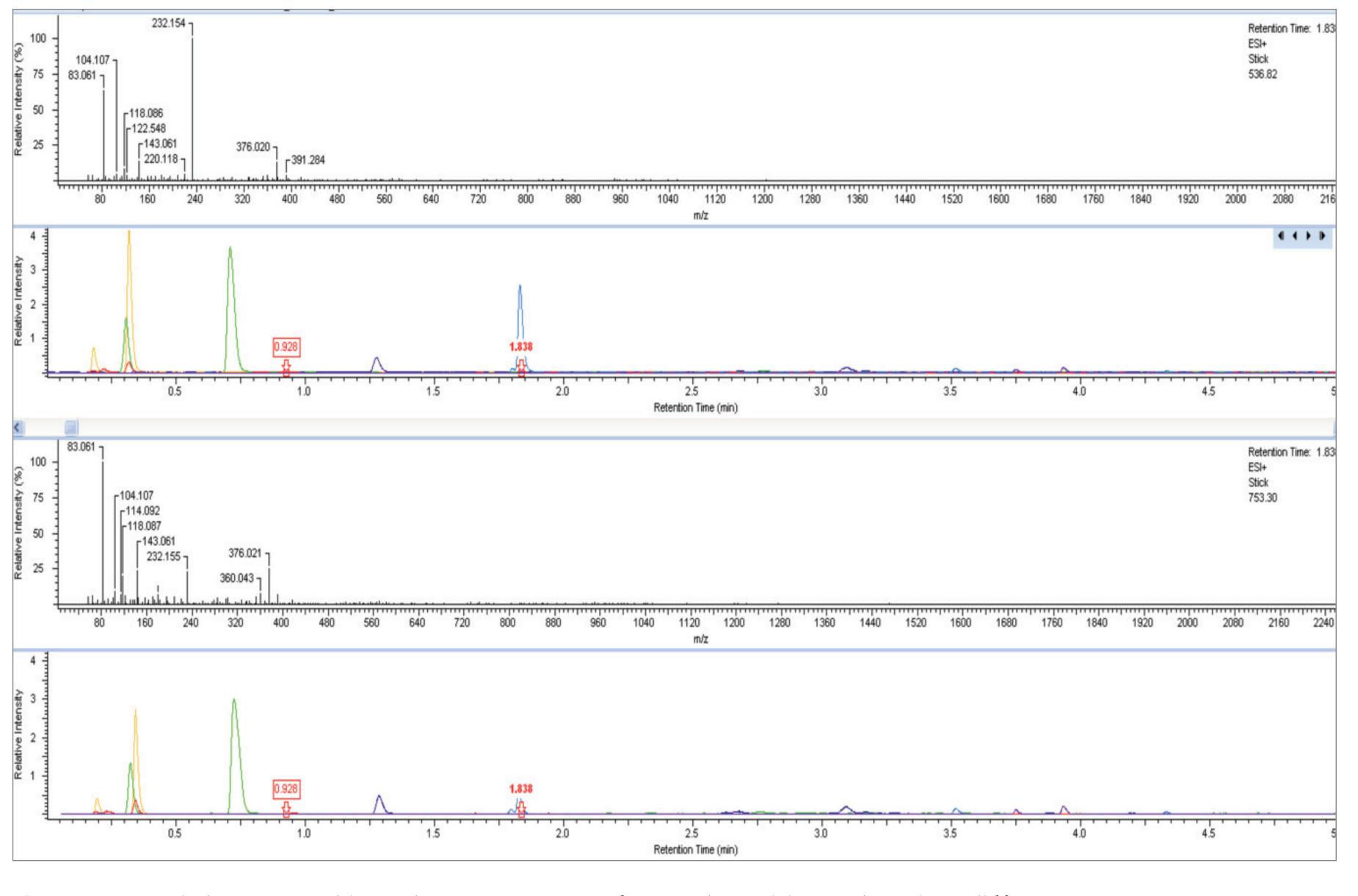


Figure 2. Partial extracted ion chromatograms for acylcarnitines showing differences between obese and lean rats.



<sup>2</sup> LECO Corporation, Saint Joseph, MI



### **RESULTS and DISCUSSION**

operating at R (FWHM) = 50,000 using external calibration and reliable relative isotope abundance. For example, the only elemental formula found within 5 ppm of the C4:0 was the correct one,  $C_{11}H_{21}NO_4$ , accurate to 4 decimal places. Additional evidence could be provided by in source collision induced dissociation (isCID). The isCID spectra provided fragment ions with accurately measured masses. For the analysis of bacterial extracts, ACD/IntelliXtract was used. The isotopic envelope of Iron compounds is challenging to detect since the first isotope is only 6% abundant relative to the A+2 isotope and A+1 is dependent solely upon the organic moeity. However, relative isotopic abundance and characteristic mass differences were extremely useful in confidently identifying some metal ion complexes in the mixtures, see Figure 1.

## CONCLUSIONS

ACD/IntelliXtract COMPARE function processing offers a means to find components that vary between samples. Amino acids and lipids were noted among the components. In the siderophore sample, certain components containing metals gave characteristic isotope patterns. The LC-HRT performance characteristics aid formula generation because of excellent mass accuracy and reliable relative isotope abundance. Further structure information can be gleaned from in source collision induced dissociation (isCID) especially since the isCID spectra provided fragment ions with accurately measured masses.

While it was possible to extract chromatograms, further optimization will be needed to detect iron siderophores reliably since the relatively low abundance of the first two isotopes compared to the third makes them challenging to detect automatically.

