

Effective Comparison of Yeast Extracts Using High Resolution GC and GC×GC-HRTOFMS

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Introduction

Yeast has been used for centuries in the production of beer, wine, and baked goods.¹ In modern times, yeast has become a “cellular factory” for the manufacture of bioethanol, lactic acid, and other valuable organic molecules.²⁻⁴ In order for this bioengineering to become realized, the metabolism of yeast must be clearly understood. Unfortunately, yeast is difficult to study due to the chemical and physical complexity of metabolites and their wide concentration range. Furthermore, the bottleneck in metabolomics research and engineering is the characterization of the diverse features present in these samples. In this study we explored the utilization of gas chromatography high resolution time-of-flight mass spectrometry (GC-HRT) and GC×GC-HRT for efficient metabolite chromatographic separation, minimization of background interference, resolution of isobaric interferences, and rapid compound identification.

Methods

Objectives

- 1) Demonstrate the utility of multidimensional chromatography and high resolution mass spectrometry for rapid and confident identification of yeast metabolites.



Fig. 1: LECO Pegasus® GC-HRT 4D.

- Speed and robustness
- Unprecedented chromatographic resolution
- EI and CI data acquisition
- High resolution deconvolution
- High quality spectral data for spectral similarity searches (NIST, Wiley, etc.)
- High resolution accurate mass measurements for fragment, molecular, and adduct ion formula determinations

- 2) Investigate the effectiveness of yeast extraction using a mixed solvent system (H₂O/CH₃OH/CHCl₃).
- 3) Compare metabolites partitioned between polar and non-polar phases.
- 4) Pinpoint key metabolites differentiating commercial dry yeast samples via statistical analysis.⁵

Samples



- 2 yeast autolysate samples (A and B); GC and GC×GC system capabilities
- Dry baker's yeast samples (1–5); Extraction investigations

Extraction

Yeast (15 to 30 mg) was extracted with 5 mL 0.5:0.5:1 H₂O/CH₃OH/CHCl₃ and filtered into 2 mL GC vials. The solvents were evaporated using a Speed Vac and residual water was removed with a lyophilizer.

Derivatization

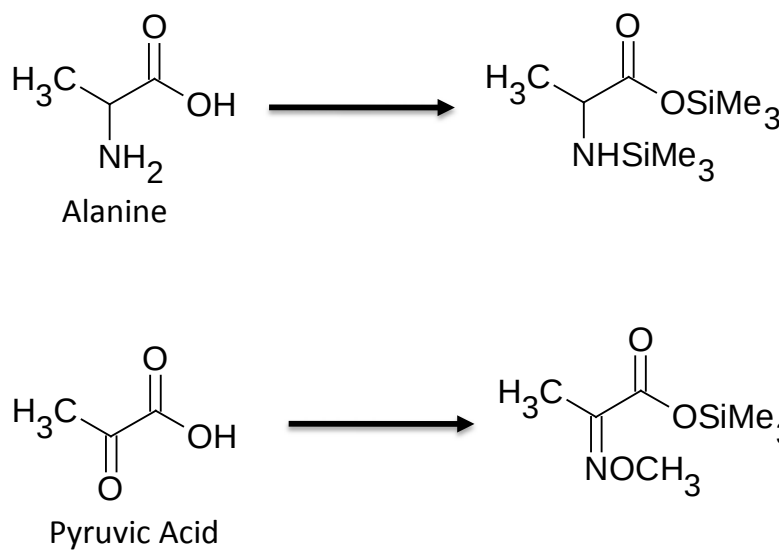


Fig. 2: Derivatization of Alanine and Pyruvic Acid.

- a) 20 µL of MEOX (20 mg/mL methoxyamine in pyridine), 1 hour at 60°C
- b) 80 µL of MSTFA, 1 hour at 60°C

Instrument Parameters

Gas Chromatograph	Agilent 7890 with MPS2 Autosampler
Injection	1 µL, split 20:1 @ 250°C
Carrier Gas	He @ 1.0 mL/min, Constant Flow
Column	Rxi-5ms, 30 m x 0.25 mm i.d. x 0.25 µm coating
Oven Program	1 min at 60°C, ramped 10°C/min to 320°C, held 10 min
Transfer Line	300°C
Mass Spectrometer	LECO Pegasus HT
Ion Source Temperature	250°C (CI: 200°C)
Mass Range	35-510 m/z (CI 60 -1200)
Acquisition Rate	10 spectra/s

Gas Chromatograph	Agilent 7890 with Dual Stage Quad Jet Modulator and MPS2 Autosampler
Injection	1 µL, split 20:1 @ 250°C
Carrier Gas	He @ 1.0 mL/min, Corrected Constant Flow
Column One	Rxi-5ms, 30 m x 0.25 mm i.d. x 0.25 µm coating
Column Two	Rxi-17SilMS, 2.0 m x 0.25 mm x 0.25 µm coating
Temperature Program	1 min at 60°C, ramped 10°C/min to 320°C, held 10 min Secondary oven maintained +5°C relative to primary oven
Modulation	3 s with temperature maintained +15°C relative to 2nd oven
Transfer Line	300°C
Mass Spectrometer	LECO Pegasus HRT
Ion Source Temperature	250°C (CI: 200°C)
Mass Range	35-510 m/z (CI 60 -1200)
Acquisition Rate	200 spectra/s

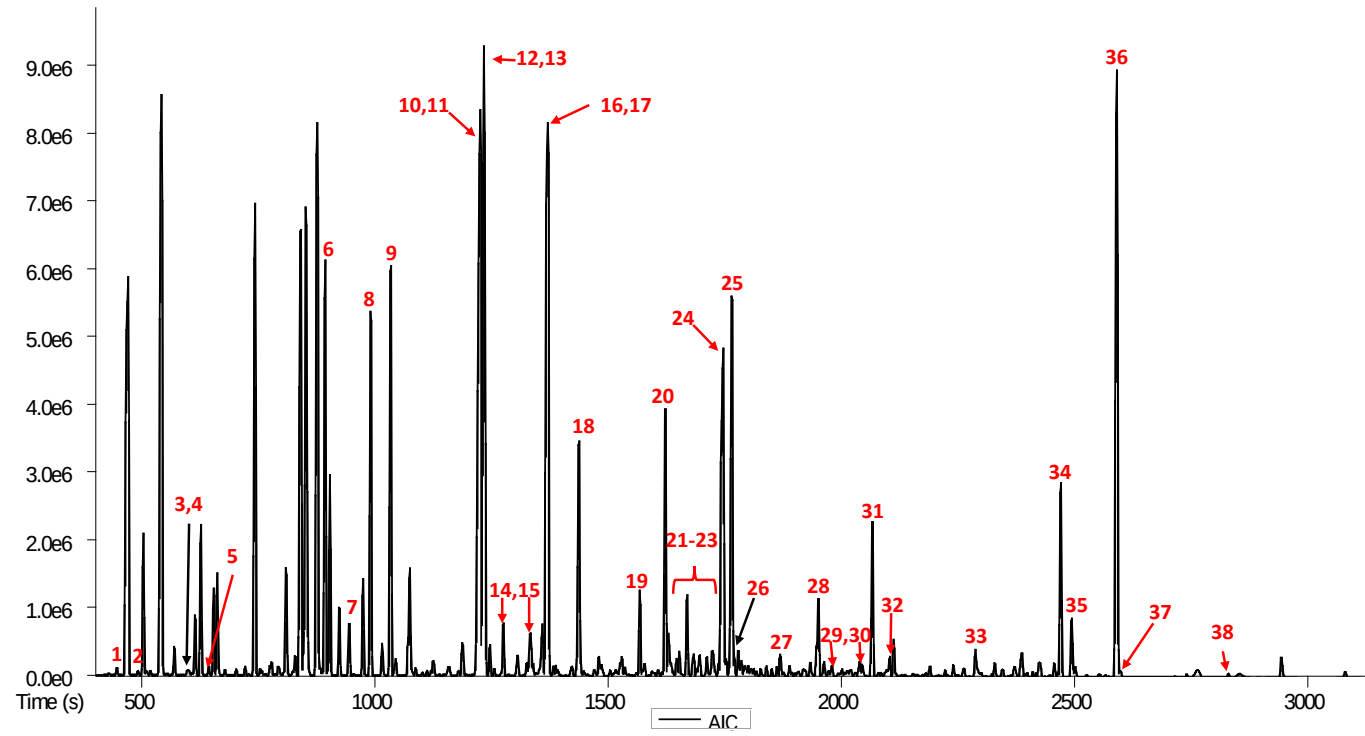


Fig. 3: Analytical Ion Chromatogram (AIC) Yeast Extract A.

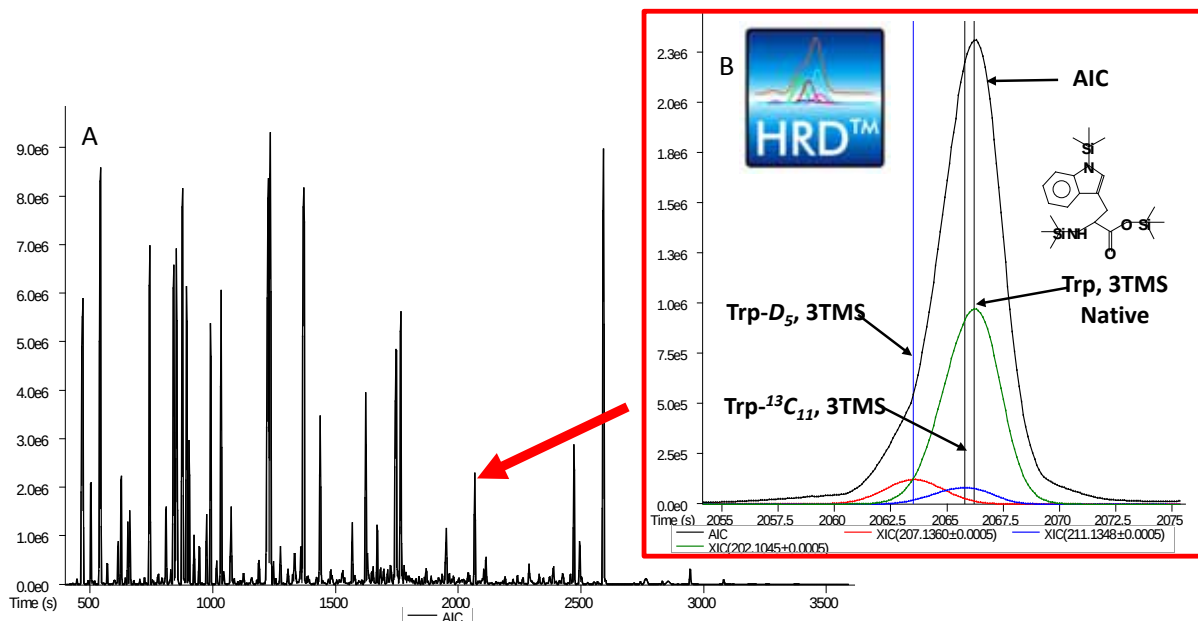


Fig. 4: AIC and Extracted Ion Chromatogram (XIC) For D₆, ¹³C₁₁, and Native Trp (3TMS).

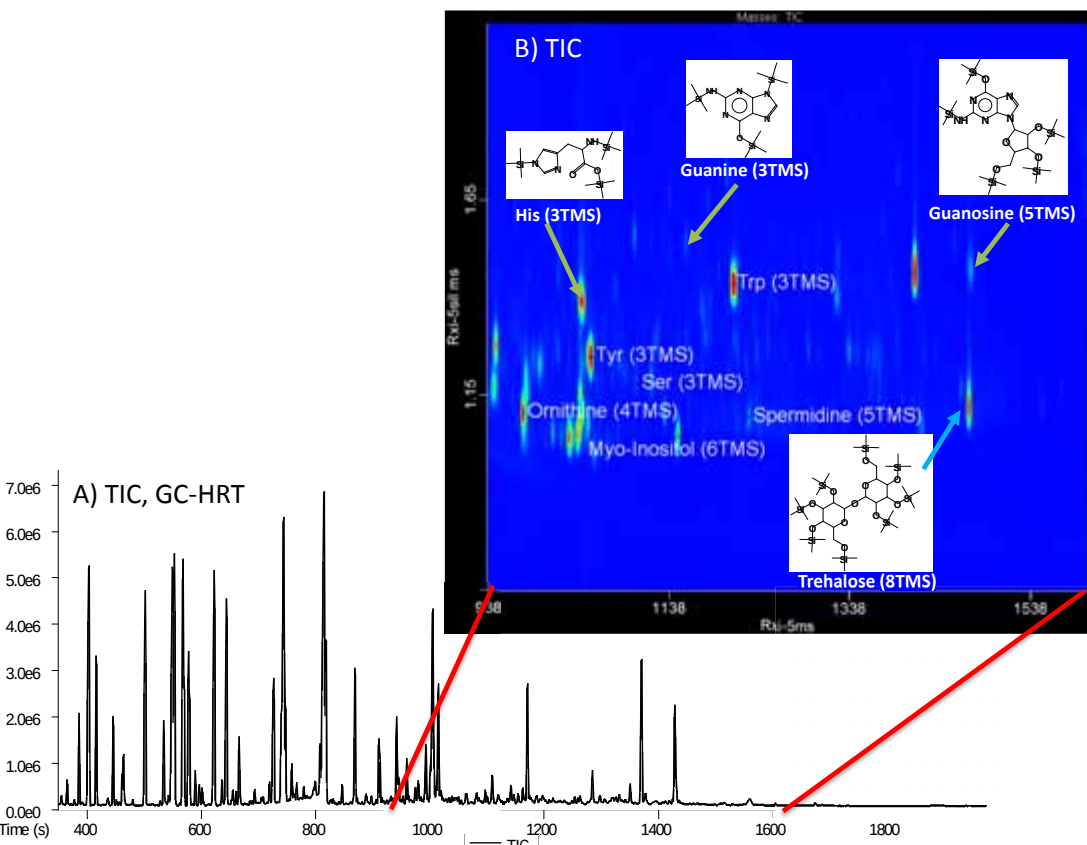


Fig. 7: A) TIC and B) GC×GC-HRT Contour Plot for Yeast Extract B.

Results: GC-HRT (Yeast A)

Table 1: Representative Compounds in Yeast Extract A.

Peak #	Name	Formula	R.T. (s)	Area	Similarity
1	Pyruvic Acid (MEOX, TMS)	C ₃ H ₄ O ₃ Si	446	113787	934
2	Glycolic Acid (2TMS)	C ₂ H ₄ O ₃ Si ₂	491	715276	920
3	2-Furancarboxylic Acid (TMS)	C ₆ H ₄ O ₃ Si	588	236557	910
4	Oxalic Acid (2TMS)	C ₂ H ₂ O ₄ Si ₂	588	582722	920
5	3-Hydroxybutyric Acid (2TMS)	C ₄ H ₆ O ₄ Si ₂	643	181101	900
6	Isolactic Acid (2TMS)	C ₃ H ₆ O ₃ Si ₂	676	807728	964
7	Glyceric Acid (2TMS)	C ₃ H ₆ O ₄ Si ₂	944	384518	811
8	Serine (2TMS)	C ₃ H ₇ NO ₃ Si ₂	990	3702912	925
9	Threonine (3TMS)	C ₄ H ₉ NO ₃ Si ₃	1031	4251682	948
10	Methionine (2TMS)	C ₄ H ₉ NO ₂ Si ₂	1232	1229718	825
11	5-oxo-Proline (2TMS)	C ₅ H ₈ NO ₃ Si ₂	1226	21048974	928
12	Aspartic Acid (2TMS)	C ₄ H ₇ NO ₄ Si ₂	1233	7342192	957
13	4-Aminobutanoic Acid (3TMS)	C ₄ H ₉ NO ₃ Si ₃	1235	4689530	924
14	Cysteine (3TMS)	C ₃ H ₇ NO ₂ Si ₂	1275	548421	961
15	Ornithine (3TMS)	C ₆ H ₁₁ NO ₃ Si ₃	1354	520005	908
16	Phenylalanine (2TMS)	C ₉ H ₉ NO ₂ Si ₂	1367	5269227	955
17	Glutamic Acid (2TMS)	C ₅ H ₉ NO ₄ Si ₂	1371	9769927	866
18	Asparagine (3TMS)	C ₄ H ₈ N ₂ O ₃ Si ₃	1437	24267014	952
19	Glycerophosphate (4TMS)	C ₃ H ₈ O ₄ PSi ₄	1567	5728912	905

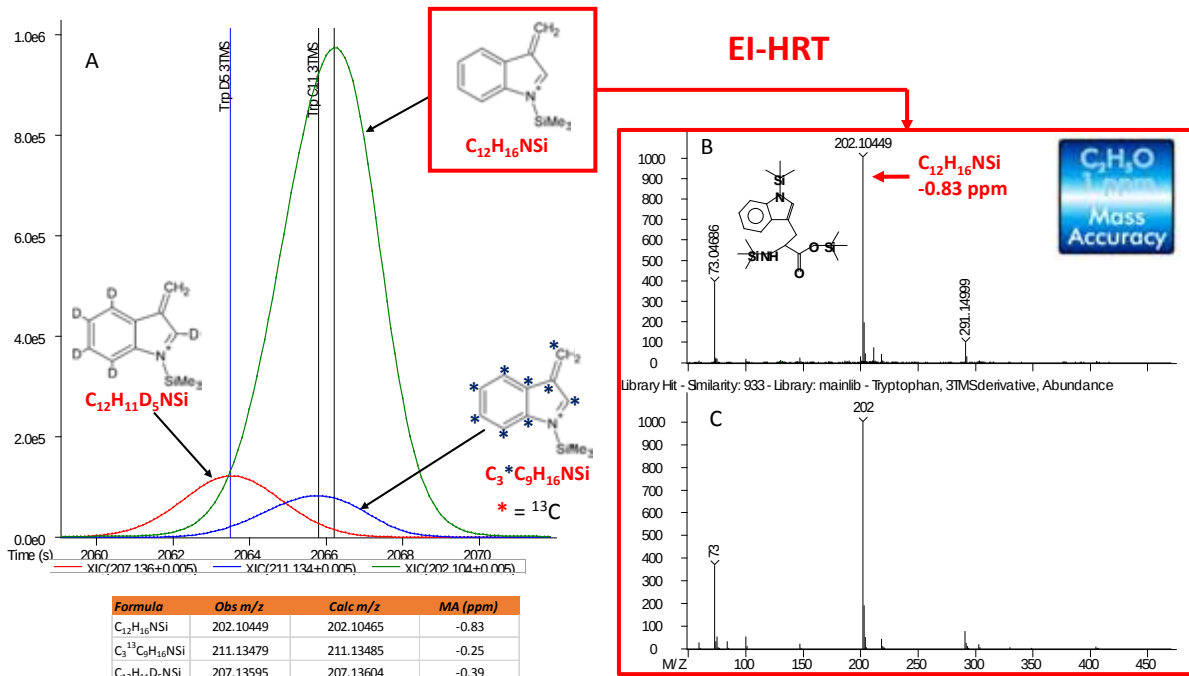


Fig. 5: A) Mass Accuracy Values for D₆, ¹³C₁₁, and Native Trp (3TMS). B) Peak True and C) Library Mass Spectra Native Trp (3TMS).

Results: GC×GC-HRT (Yeast B)

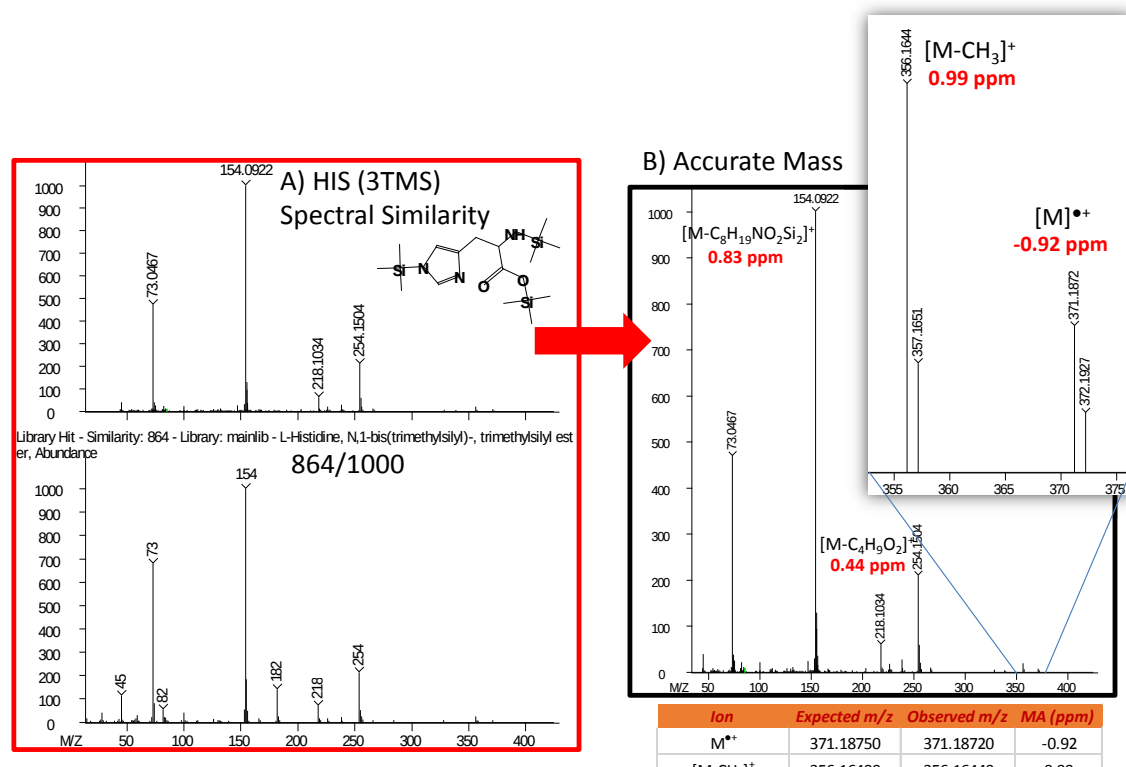


Fig. 8: A) Peak True and Library Mass Spectra for His (3TMS). B) Expansion of Peak True Mass Spectrum and Mass Accuracy Values for His (3TMS).

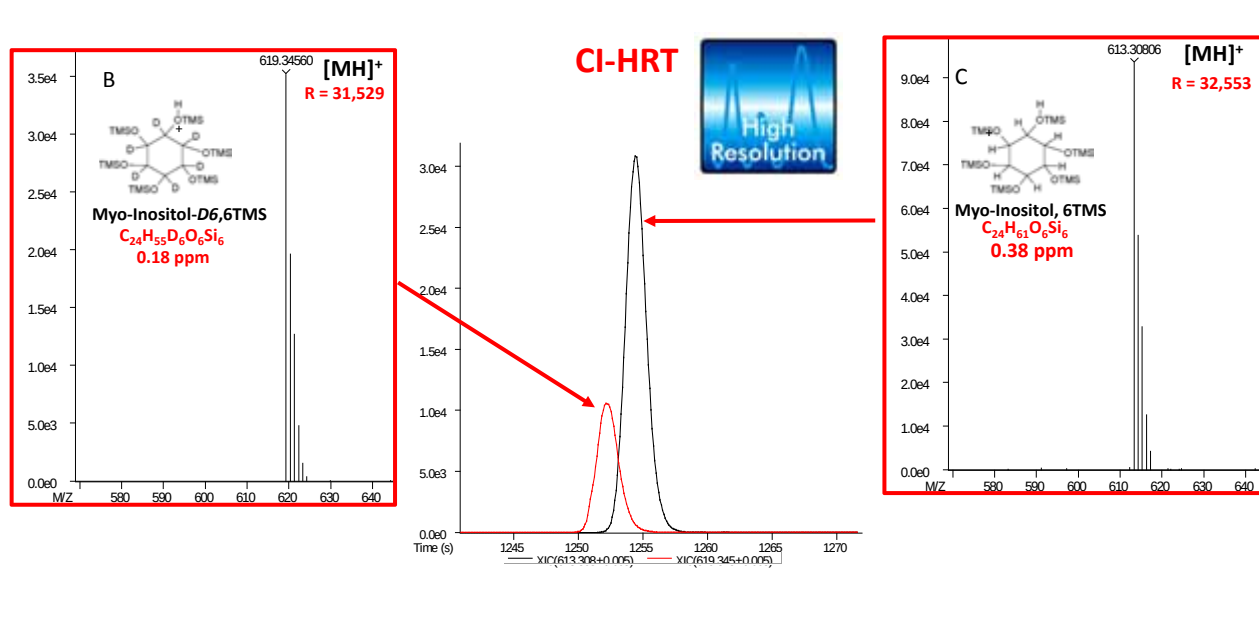


Fig. 6: A) XIC for D₆ and Native myo-Inositol. CI-HRT Mass Spectra for B) D₆, and C) Native myo-Inositol.

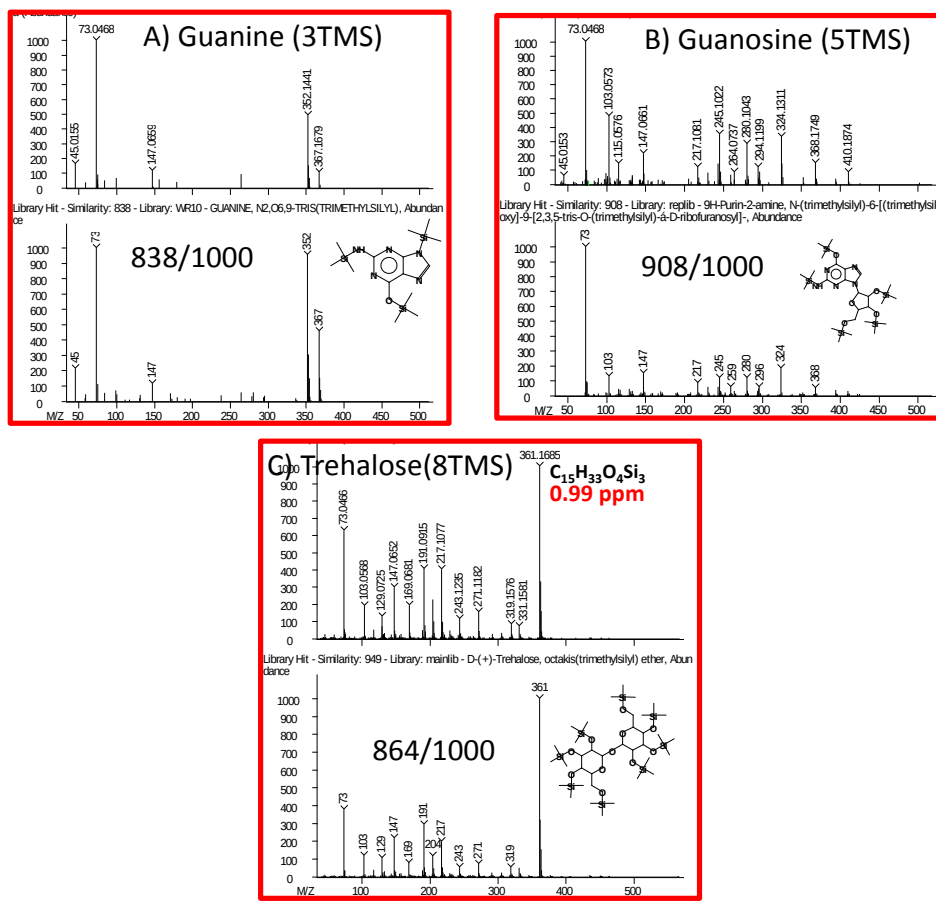


Fig. 9: Peak True and Library Mass Spectra for A) Guanine (3TMS), B) Guanosine (5TMS), and C) Trehalose (8TMS).

Results: GC-HRT Comparison of Dry Yeast Extracts 1–5



Fig. 10: PCA Plot (Fold Change > 1.5, P < 0.05) for Baker's Yeast Samples Extracted Using a Two Phase Solvent System (T = Aqueous, B = Organic). Sample 5 Did Not Contain Additional Additives (e.g., ascorbic acid, sorbitan monostearate).

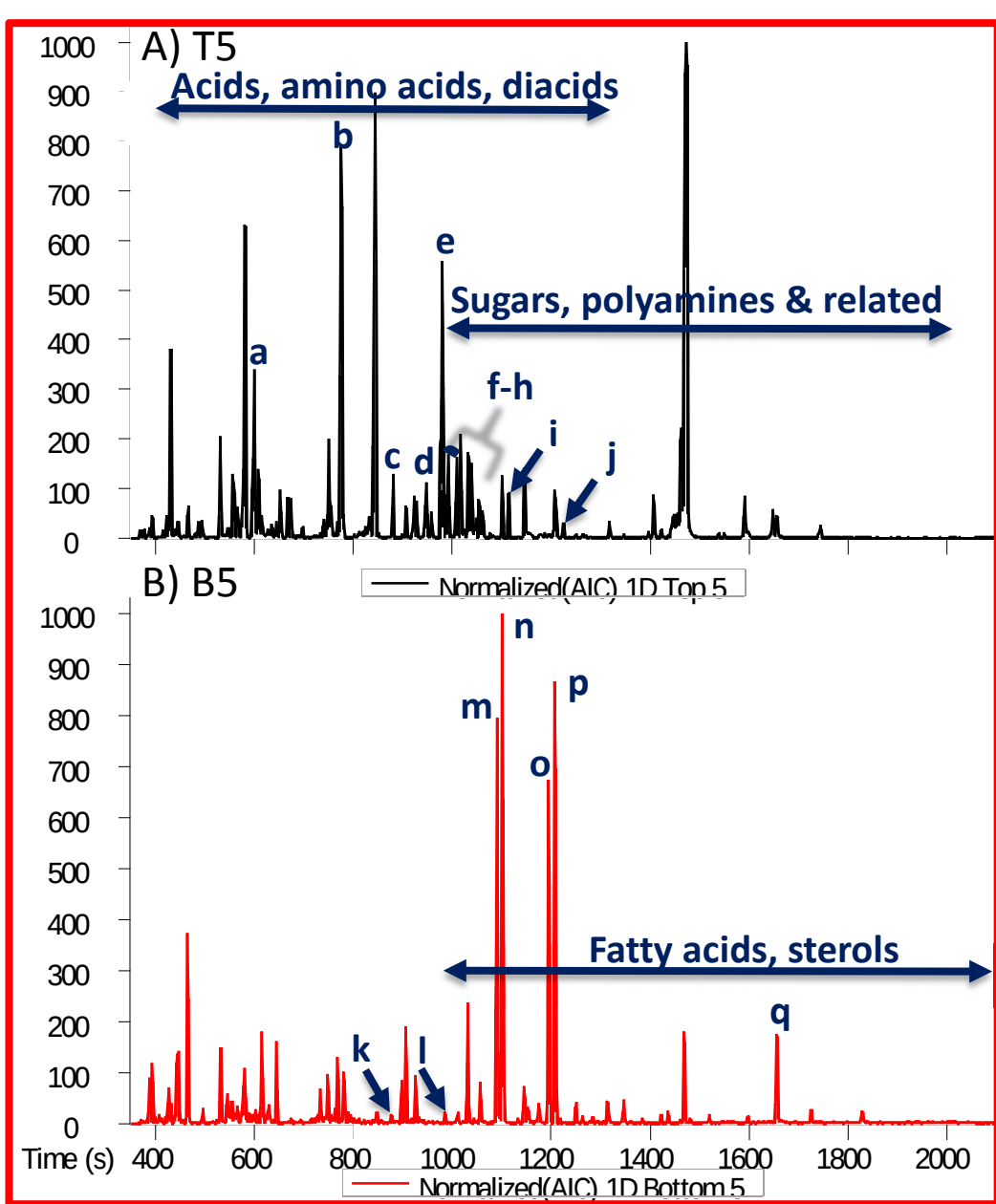


Fig. 11: Top (Aqueous) and Bottom (Organic) AICs for Yeast Extract 50

Tables 2 and 3: Yeast Extract 5 Representative Compounds Extracted Into Top (Aqueous) and Bottom (Organic) Phases.

Top (Aqueous)					
Peak	Name	R.T. (s)	Formula	Area	Similarity
a	Proline (2TMS)	600	C ₅ H ₁₀ NO ₂ Si ₂	14228968	925
b	Glutamic Acid (3TMS)	844	C ₆ H ₁₂ NO ₄ Si ₃	19581921	918
c	Asparagine (3TMS)	880	C ₄ H ₈ N ₂ O ₃ Si ₃	3150080	784
d	Glutamine (3TMS)	946	C ₆ H ₁₄ N ₂ O ₃ Si ₃	939026	835
e	Citric Acid (4TMS)	980	C ₆ H ₈ O ₇ Si ₄	10913589	839
f	Fructose (MEOX, 5TMS)	1009	C ₁₂ H ₂₂ NO ₆ Si ₅	3549185	815
g	Mannose (MEOX, 5TMS)	1035	C ₁₂ H ₂₂ NO ₆ Si ₅	3311292	933
h	Galactose (MEOX, 5TMS)	1046	C ₁₂ H ₂₂ NO ₆ Si ₅	958196	919
i	Cys-Gly (3TMS)	1114	C ₄ H ₈ N ₂ O ₃ Si ₃	1755190	924
j	Spermidine (5TMS)	1225	C ₁₃ H ₃₀ N ₄ Si ₅	610033	875

Bottom (Organic)					
Peak	Name	R.T. (s)	Formula	Area	Similarity
k	Lauric Acid (TMS)	859	C ₁₂ H ₂₄ O ₂ Si	21301	768
l	Myristic Acid (TMS)	985	C ₁₄ H ₂₈ O ₂ Si	42326	771
m	Palmitoleic Acid (TMS)	1090	C ₁₆ H ₃₀ O ₂ Si	1100651	920
n	Palmitic Acid (TMS)*	1101	C ₁₆ H ₃₂ O ₂ Si	2764956	916
o	(E)-Oleic Acid (TMS)	1195	C ₁₈ H ₃₄ O ₂ Si	865860	924
p	Stearic Acid (TMS)*	1207	C ₁₈ H ₃₆ O ₂ Si	1925246	898
q	Ergosterol**	1656	C ₂₈ H ₄₆ O ₂ Si	250513	849

Results: GC and GC×GC-HRT Dry Yeast Extract #5

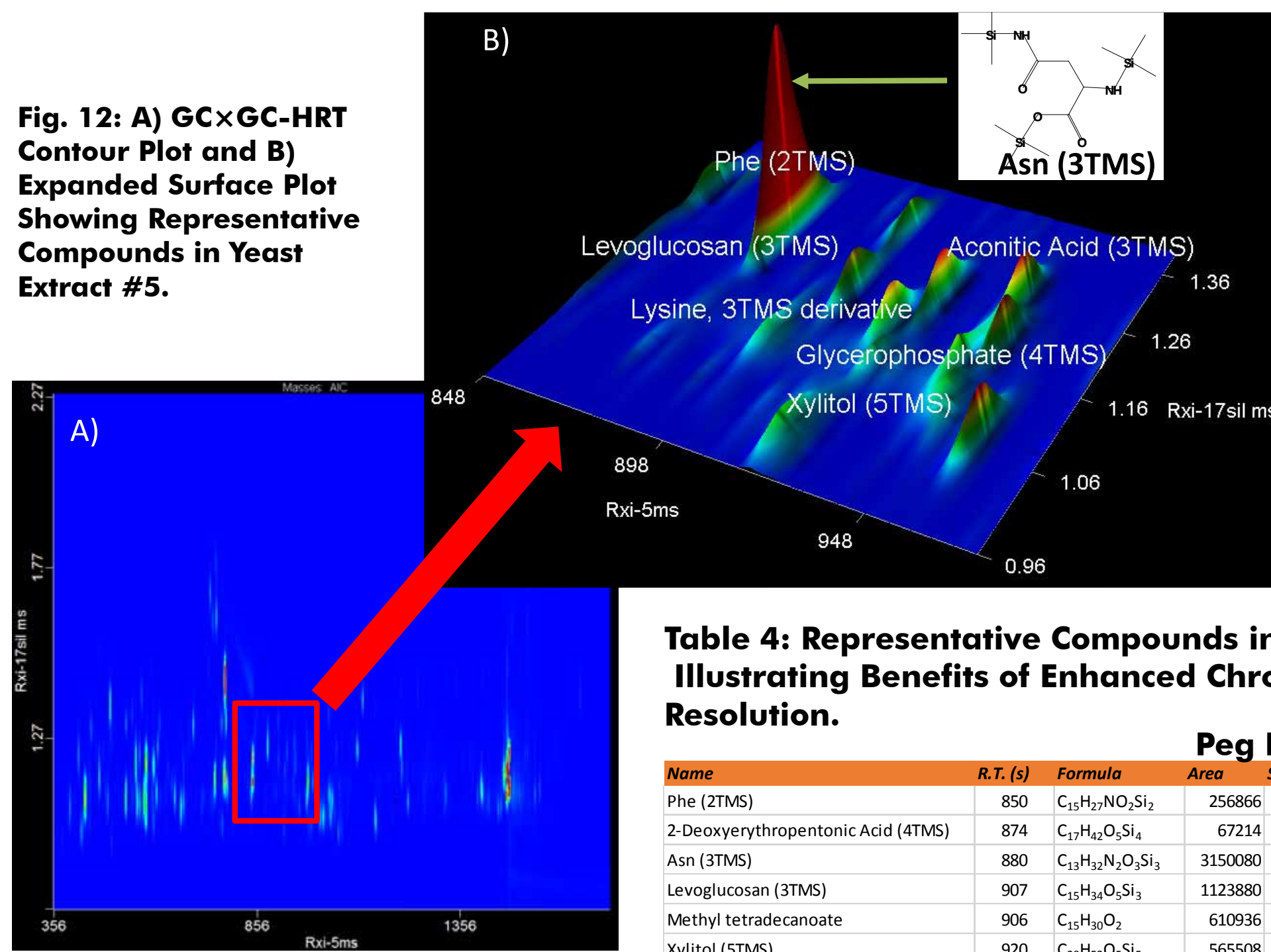


Fig. 12: A) GC×GC-HRT Contour Plot and B) Expanded Surface Plot Showing Representative Compounds in Yeast Extract #5.

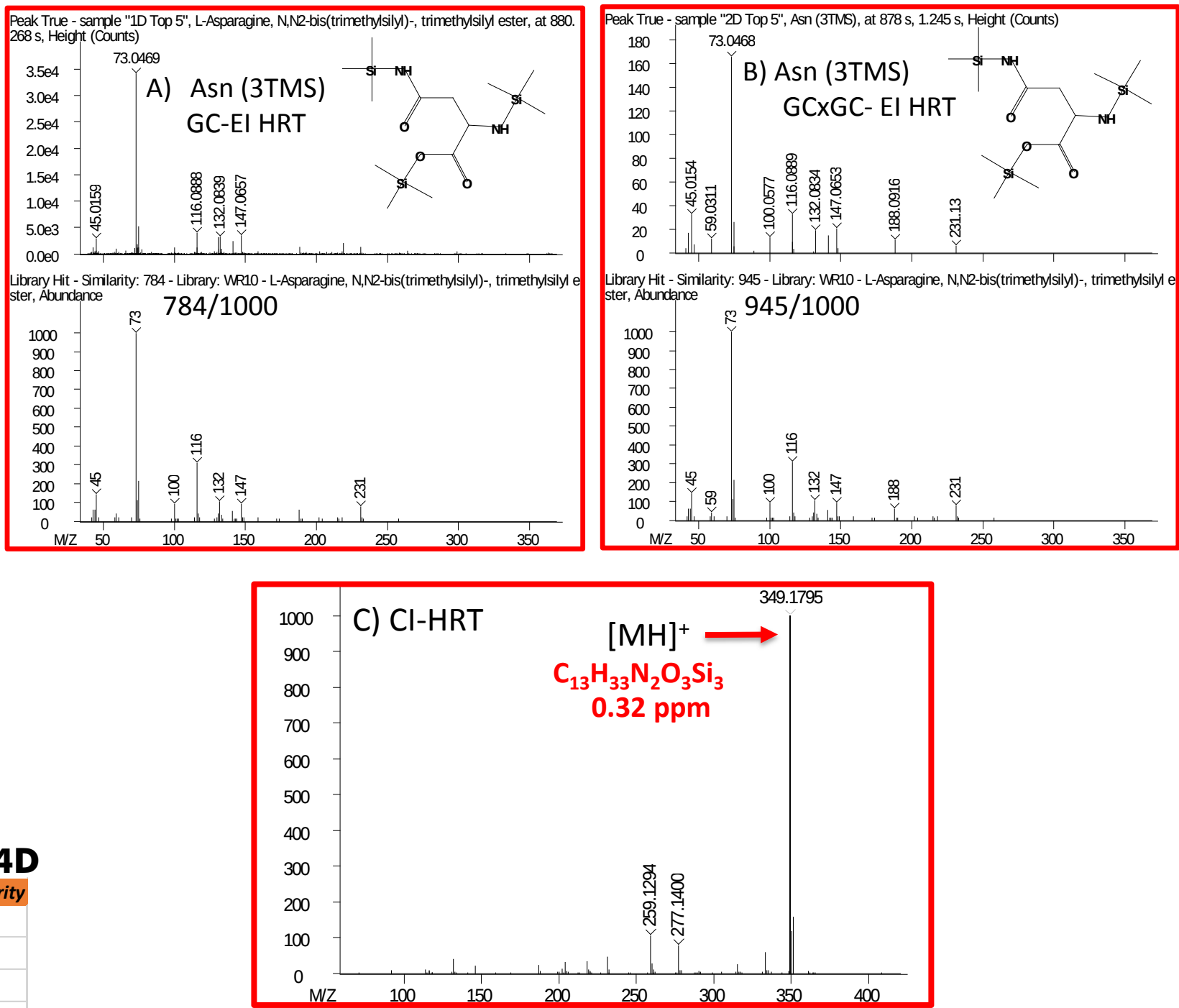


Fig. 13: Peak True and Library Mass Spectral Data for Asn (3TMS) Acquired Using A) Pegasus GC-HRT and B) Pegasus GC-HRT 4D instruments. C) CI-HRT data for Asn (3TMS).

Conclusions

The Pegasus GC-HRT and GC-HRT 4D instruments are effective tools for the comprehensive analysis of yeast extracts.

- Enhanced chromatographic resolution of the HRT 4D system facilitated compound characterization by improving spectral similarity scores
 - High quality data resulted in excellent spectral similarity scores when compared to large, well-established databases (NIST, Wiley)
 - High resolution, accurate mass data was crucial for determination of fragment, molecular, and adduct formulas
- A mixed solvent system was used for extraction of yeast metabolites.
- Polar protic solvents such as methanol and water are effective solvents for extraction of small polar and/or ionic metabolites
 - Less polar solvents (e.g., chloroform) are superior for the extraction of fatty acids, sterols, and other relatively non-polar materials

References

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