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# Investigating the Effects of Commercial Antimicrobial Agents on Human Corneal Epithelial Cell Membranes lan J. Horner, Jerod J. Hurst, Nadine D. Kraut, Alyssa A. Rook, Crystal M. Collado, G. Ekin Atilla-Gokcumen, and Frank V. Bright\*

#### Abstract

Constituents in multipurpose contact lens solution (MPS) have been suggested to cause corneal injury. To explore this issue we created an *in-vitro* liposome-based model of the corneal epithelial surface and we assessed the interactions of polyhexamethylene biguanide (PHMB) and polyquaternium-1 (PQ-1) on the biomembrane by using spectroscopy, dynamic light scattering (DLS), and fluorescence chromatography/mass spectrometry (LC-MS). Fluorescence assessed the membrane surface polarity and stability through the temperature-dependent generalized polarization (GP) the gel-to-liquid transition temperature  $(T_m)$  and the associated van't Hoff enthalpy  $(\Delta H_{VH})$ . DLS evaluated liposome-liposome aggregation. LC-MS determined the composition of any precipitates that formed. PHMB increased  $T_m$ , phospholipid cooperativity, and GP. In contrast, PQ-1 did not change  $T_m$  or phospholipid cooperativity, but it decreased GP. PQ-1 alone lead to liposome-liposome aggregation. The aggregates exhibited a liposome composition equivalent to the liposome prior to the addition of PQ-1.

### Background

In the eye care industry, MPSs are used extensively for soft contact lens disinfection, cleaning and storage. Commercial MPSs are composed of a buffer system, at least one antimicrobial agent (AA), and other additives such as surfactants or humectants that are used to provide lens comfort and improve performance. PHMB and PQ-1 are among the most widely used AAs.<sup>1</sup>

After a contact lens is soaked, removed from a MPS and applied to the ocular surface, AAs begin to desorb from the contact lens at a rate that depends on the agent and the contact lens material.<sup>2</sup> Within the human pre-corneal tear film, PHMB and PQ-1 exist as polycations.<sup>3</sup> Thus, strong evidence for ion-ion interactions between PHMB and PQ-1 and oppositely charged functionalities present at cell surfaces is well established.<sup>4</sup>

# Antimicrobial agent chemical structures: Polyquaternium-1 (PQ-1) Polyhexamethylene biguanide (PHMB) Major phospholipids of the human corneal epithelial cell membrane<sup>5</sup>: N-palmitoyl-D-erythro-sphingosylphosphorylcholine 1,2-dipalmitoyl-sn-glycero-3-phosphoethanolamine CH<sub>3</sub>(CH<sub>2</sub>)<sub>13</sub>CH<sub>2</sub>, он 1,2-dipalmitoyl-*sn*-glycero-3-phospho-(1'-myo-inositol 1,2-dipalmitoyl-sn-glycero-3-phosphocholine (DPPC) 1,2-dipalmitoyl-sn-glycero-3-phospho-L-serine $N_3$ (DIIO) Laurdan Acyl chains tilted with respect to vesicle normal Lipid vesicle 24 Å

#### Theory

The fluorescent reporter, Laurdan, is sensitive to its local microenvironment. Laurdan displays a large emission red shift in polar solvents in comparison to non-polar solvents. To quantify the Laurdan spectral shifts we computed the generalized polarization (GP)<sup>6</sup>:

$$GP = (I_{440} - I_{490}) / (I_{440} + I_{490})$$
(1)

It is well-known that phospholipid bilayers undergo reversible temperature-dependent gel-to-liquid phase transitions. This transition is described by a characteristic  $T_m$  (defined by the mid-point in the GP vs. T profile<sup>6</sup>) and by an associated van't Hoff enthalpy ( $\Delta H_{VH}$ ) that reflects subunit (phospholipid) cooperativity during the gel-to-liquid phase transition.

$$\Delta H_{\rm VH} = 4RT_{\rm m}^{2}(\delta\alpha/\delta T)_{\rm Tm} \qquad (2)$$

$$\alpha = [(GP)_{max} - (GP)_{T}] / (\Delta GP)_{\Delta T} \quad (3)$$

where

