

# Recycling of CO<sub>2</sub> on a Supercritical Fluid Chromatography (SFC) Pilot Unit



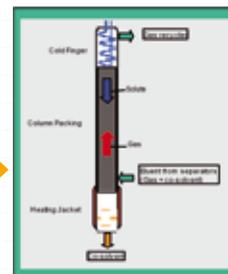
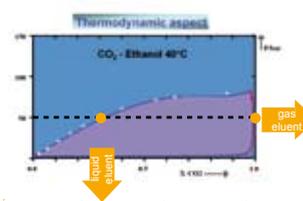
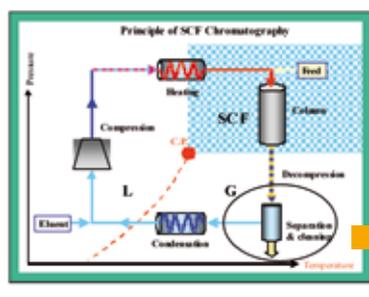
Wieslaw Majewski, François Romelaer, Olivier Ludemann-Hombourger

Novasep SAS - Site Eiffel, 81 Bld de la Moselle - BP50 - 54340 Pompey - France

## Introduction

Except for analytical and semi-prep instruments, eluent recycling after solute separation should be an ordinary feature of any supercritical fluid extraction and chromatographic equipment. But it should be stressed, that gaseous eluent from separators always contains some traces of solute and co-solvent,

entrained as particles or vapor. Hence, in order to recycle the CO<sub>2</sub> and reuse it, the gas leaving the separators must be further purified. This purification step must provide clean CO<sub>2</sub> at the same pressure as a fresh fluid delivered from the supply unit (40 to 50 bar).



## Practical Aspects

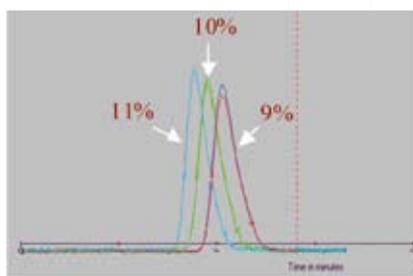
The purification of CO<sub>2</sub> is achieved in a gas scrubber. The system removes co-solvent and solute traces entrained from cyclonic separators. The remaining co-solvent dissolved in gaseous CO<sub>2</sub> is very low (< 0.3% under typical conditions) and is stable. Therefore, it has no influence on the chromatographic separation.

## Impact on the Chromatography Separation

In SFC, even a small change in eluent composition typically conducts to a visible change in peak retention time.

Preparative chromatography involve multiple injections (10's to 1000's) and automatic collections of purified products. Therefore, any instability of the retention time would be detrimental to the purity and recovery yield. The performance of the recycling was demonstrated as shown in the figure on the right.

**Conclusion:** A perfect reproducibility is obtained at each CO<sub>2</sub>/co-solvent composition tested.



Overlay of 3 pairs of chromatograms obtained with 3 co-solvent contents



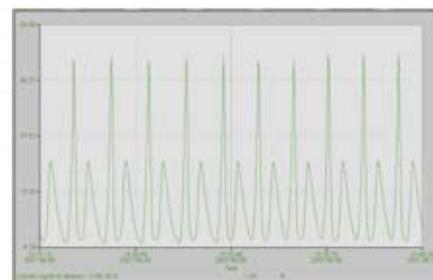
SuperSep™: A compact SFC unit  
Capacity: kg(s) of product/day

## Impact on the solvent consumption

System Type	Column Size	HPLC 80 mm I.D.	SFC 50 mm I.D.	SFC 50 mm I.D.
Productivity (kg product/day/kg CSP)		0.240	2.2	2.2
Organic Solvent (L/kg of feed)		3100 (n-heptane/IPA 98/2)	352 (IPA+DMEA)	352 (IPA+DMEA)
CO <sub>2</sub> consumption (kg/kg of feed)		NA	1700 <b>(Without CO<sub>2</sub> recycling)</b>	170 <b>(With CO<sub>2</sub> recycling)</b>
Productivity (time to purify 1 kg)		100 h	36 h	36 h
Manufacturing cost comparison (on hundred basis)		100	49	8.8

Separation of cyclopenta [b] Indol Amino derivative (B.Bonnier -R&D Sr Scientist, Eli Lilly, Belgium)

Recycling the CO<sub>2</sub> reduces the required quantities of fresh CO<sub>2</sub> necessary for semi-prep to preparative separation by 10 to 20 times. This can be easily translated by the reduction of the dimension of the CO<sub>2</sub> furniture, and therefore **the reduction of the capital expenditure** and the logistics involved.



Series of injections of chiral API

## Conclusion

SFC is a powerful separation tool to all chemists. With CO<sub>2</sub> as eluent, SFC separates fast, and well.

**The advantages of the technique are:**

- High productivity
- Low amounts of organic solvents (5 to 20 times less than HPLC)
- Concentrated purified products (Size reduction of evaporators)

**What does CO<sub>2</sub> recycling bring to your SFC?**

- Reduction of CO<sub>2</sub> storage dimensions
- Decrease of global investments
- Reduction of operating costs
- Reduction of the risks associated with large CO<sub>2</sub> volumes